Name:_____

7.2 and 7.3 Review

Complete the Following Chart:

,

| Standard Notation | # of Protons | # of Neutrons | # of Electrons | Mass Number | Atomic Number |
|----------------------|-----------------|------------------|-------------------|----------------|------------------|
| | | 45 | 36 | | 35 |
| | 26 | | 23 | | |
| Zn | | | | | |
| Ι | | | | | |
| | 11 | | 11 | 23 | |
| | | 8 | 6 | 14 | |
| N ³⁻ | | | | | |
| | | 15 | 15 | | 15 |
| | 3 | 4 | 2 | | |
| Al ³⁺ | | | | | |

1. How many electron orbitals does Strontium have? _____

2. Where are the positive charges in an atom found? _____

3. Where are the negative charges in an atom found? ______

4. Name the six families of elements that we have discussed: _____,

5. Which group of elements tend to form positive ions? _____

6. Which group of elements tend to form negative ions? _____

7. Identify the follow information: atomic number, most common ion charge, symbol of the element, name of element, and atomic mass